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ZERO FIELD NMR

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ABSTRACT

In polycrystalline samples, NMR 'powder spectra' are broad and much structural information is lost due to the orientational disorder. In this letter we describe Fourier transform NMR in zero magnetic field. With no preferred direction in space, all crystallites contribute equivalently and resolved dipolar splittings can be interpreted directly in terms of internuclear distances. This opens the possibility of molecular structure determination without the need for single crystals or oriented samples.

Techniques for studying molecular structure, such as x-ray diffraction and nuclear magnetic resonance (NMR), often require the use of single crystals (1) or oriented samples (2). NMR, for example, is sensitive to interatomic distances and angles through spectral splittings caused by internuclear dipole-dipole couplings in high magnetic field. Where only polycrystalline powders or amorphous materials are available, the distribution of molecular orientations (and thus internuclear vectors) with respect to the direction defined by the x-ray beam or by the magnetic field, gives rise to a 'powder pattern' in which most structural information is lost (3).

Consider the notion of NMR with zero magnetic field, i.e. spectroscopy of pure dipole-dipole interactions. Without an externally imposed direction in space, all orientations are equivalent and the spectrum of a powder should be 'crystal-like' with all equivalent molecules yielding identical splittings. We present in this letter a novel experiment, Fourier Transform Zero Field NMR, which provides just such spectra, opening the way for crystallography and molecular structure determination in polycrystalline or disordered materials.

Figure 1 illustrates the idea for the simple case of two coupled proton spins, $(Ba(ClO_3)_2 \cdot H_2O)$, showing increased resolution due to the removal of orientational broadening in zero field. Figure 1(a) is the normal high field powder spectrum. Figure 1(b) is a normal high field single crystal spectrum where the doublet splitting depends on crystal orientation with respect to the field. Figure 1(c) is the zero field powder spectrum in which the splittings assume only the largest values of the powder pattern in 1(a), resulting from the untruncated dipolar

Hamiltonian, HD.

While zero field spectroscopy is known and widely used for pure nuclear quadrupole resonance $^{(4)}$, it is primarily done in the frequency domain (the analog of continuous wave NMR) and the signals are prohibitively small except for large quadrupole frequencies. Aside from unique circumstances $^{(5)}$, it is inapplicable to NMR. For low sensitivity quadrupolar nuclei, field cycling with double resonance has been used $^{(6,7)}$. Our method employs the field cycling idea in a novel fashion and is illustrated in Figure 2. The sample is polarized in a large magnetic field B_0 , which is quenched in two steps. The second is sudden and initiates the coherent evolution period subject to H^D . At time t_1 , the field is switched back and the sample returned to B_0 for detection of a high field signal $S(t_1)$. Fourier tranformation of $S(t_1)$ results in a zero field spectrum with the full sensitivity of high field detection.

To understand the type of spectrum expected for a powder in zero field, consider an orientation dependent transformation, T, which transforms between the laboratory frame (with z along B_0), and a crystallite frame which is fixed relative to the molecules and independent of crystallite orientation in the laboratory. If the initial density operator in zero field (i.e. the one prepared in B_0) is $\rho(0)$, then at t_1 (ignoring spin lattice relaxation):

$$\langle \rho(t_1) \rangle = \langle T^{-1} \exp(-iH^D t_1) T \rho(0) T^{-1} \exp(iH^D t_1) T \rangle$$
 (1)

where <> designates an average over T for the distribution of crystallite orientations in the laboratory. While all frequencies are orientation independent, line intensities are an average over T. For the case of

normal high field Zeeman polarization ($\rho(0) = I_z$) and the simplest case of two coupled spins (i and j), evaluation of (1) yields for the zero-field free induction decay:

$$S(t_1) = \frac{Tr\{I_z \rho(t_1)\}}{Tr\{I_z^2\}} = \frac{1}{3} + \frac{2}{3} \cos \omega_{ij}^{D} t_1$$

where

$$\omega_{ij}^{D} = \frac{3\gamma^{2}h}{2\pi r_{ij}^{3}}$$

and r_{ij} is the distance between i and j. Thus we expect the Fourier transform spectrum to contain one line at 0 frequency and one each at $\pm \omega_{ij}^D$, all of equal intensity, which is the main observation of Figure 1(c). The additional features of the latter spectrum $^{(8)}$ arise from couplings between water molecules and are an indirect measure of these distances as well. The calculated proton-proton distance is 1.60Å which is identical to the results of single crystal NMR studies $^{(9)}$. The small discrepancy from neutron diffraction data $^{(10)}$ is a measure of librational motion which partially averages the observed coupling.

In larger spin systems, the spectrum will rapidly become more complex. An appealing idea is to then improve resolution by isotopic dilution of the protons in a deuterated host. $^{1}\text{H}-^{1}\text{H}$ couplings could be measured without the need for ^{2}D -decoupling $^{(11)}$ since $^{1}\text{H}-^{2}\text{D}$ dipolar couplings are effectively quenched in zero field by the non-zero asymmetry parameter of the ^{2}D electric field gradient $^{(12)}$, as has been seen in the pure NQR spectra of deuterium $^{(7)}$. As a preliminary example of this, Figure 3 shows

spectra of powdered dimethylterephthalate diluted to 16% in its perdeuterated analog. Whereas the normal high field spectrum is essentially featureless, the zero field spectrum yields resolved splittings which arise from couplings within the methyl group and between ortho aromatic protons.

We conclude with some comments on possible applications and extensions of zero field NMR.

- 1. The experiments described in this letter began with an initial Zeeman state $\rho(0) = I_z$. States of multiple spin or n-quantum correlation (13,14) would yield different zero field spectra, with additional information on dipolar couplings and structure.
- 2. The best resolved and most easily interpreted zero field spectra arise from small sets of coupled spins. One approach is to work with randomly protonated materials in a perdeuterated host. Alternatively, one could observe naturally dilute spin species such as $^{13}\text{C}-^{13}\text{C}$ (yielding backbone structures in organic molecules), $^{29}\text{Si}-^{29}\text{Si}$ or $^{1}\text{H}-^{29}\text{Si}$ (in silicates or semiconductors), or $^{1}\text{H}-^{31}\text{P}$ (in biological systems). Resolution can be improved by multiple quantum techniques $^{(14)}$ which eliminate the background signal from uncoupled spins $^{(15)}$. Isolation may also be achieved by dilution in argon matrices $^{(16)}$, with a wide range of temperatures and relaxation parameters.
- 3. By combining the zero field experiment with magic angle spinning $^{(17)}$ in a full 2-dimensional experiment $^{(18)}$ correlations would be obtained between dipolar couplings and chemical shifts. For this type of experiment the zero field should be replaced with a field matched to the sample rotation frequency $^{(19)}$.
 - 4. In cases where small quadrupolar coupling constants make direct

low frequency detection difficult, this technique could also be used to obtain the Fourier transform pure quadrupole resonance spectrum. This would be particularly valuable when no abundant high γ nuclei are available for double resonance $^{(6,7)}$, for example 23 Na and 27 Al in aluminosilicates.

In summary, we have introduced a novel zero field NMR experiment which yields dipolar couplings in powders with high resolution and sensitivity. The spectra are devoid of orientational broadening and open the possibility of structure determination in a wide range of materials without the need for single crystals or oriented samples.

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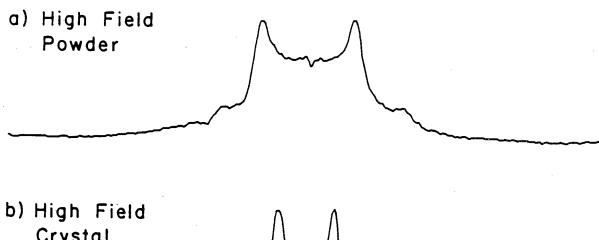
- Figure 1. Proton NMR spectra of barium chlorate hydrate $(Ba(10_3)_2 \cdot H_2 0)$. All $H_2 0$ proton-proton vectors are coparallel in the unit cell.
- a) High field powder spectrum showing normal broadened Pake doublet.
- b) High field single crystal spectrum. The sample is oriented in an arbitrary direction. The observed splitting depends on both internuclear distance and orientation.
- c) Zero field powder spectrum. The major features are those predicted by the simple treatment given in the text for a pair of coupled spins-1/2. The observed splitting is a direct, orientation independent measure of the internuclear distance. The central peak arises from spins in crystallite orientations which did not evolve in zero field.
- Figure 2. Schematic diagram of the field cycling apparatus and the time dependent effective field at the sample. The sample is moved adiabatically ($^{\sim}100$ ms) from the bore of the superconducting magnet ($^{\circ}B_0$ 4.2 Tesla), to a position $^{\circ}75$ cm below where the fringe field due to $^{\circ}B_0$ is precisely cancelled by $^{\circ}B_1$. At time $^{\circ}t_1$ = 0 coil $^{\circ}B_2$ (.01 Tesla) is switched off ($^{\circ}1\mu$ s; residual field less than $^{\circ}10^{-5}$ Tesla), and evolution under the zero field Hamiltonian is initiated. Coherent evolution is terminated by reapplying $^{\circ}B_2$, followed by transit back to the bore of $^{\circ}B_0$ where the magnetization is sampled. The entire procedure is repeated for regularly incremented values of $^{\circ}t_1$. Fourier transformation of the resulting free induction decay results in the zero field spectrum.
- Figure 3. NMR spectra of dimethylterephthalate powder. The sample is 16% dimethylterephthalate- h_{10} in dimethylterephthalate- d_{10} .

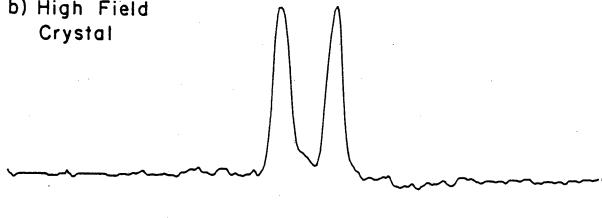
a). High field powder spectrum. b). Zero field powder spectrum. The outside pair of lines are those expected from a rapidly spinning methyl group where the interproton distances are 1.83Å. The interior pair of lines arise from the ring pairs of protons separated by 2.49\AA .

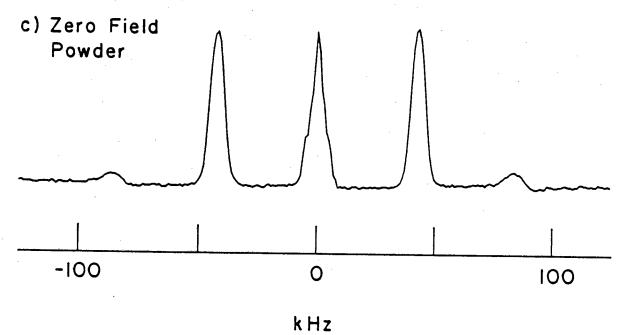
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- (8) Note the structure on the central line and the small peaks at $\pm 2\omega_{1,j}^{D}$
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Figure 1

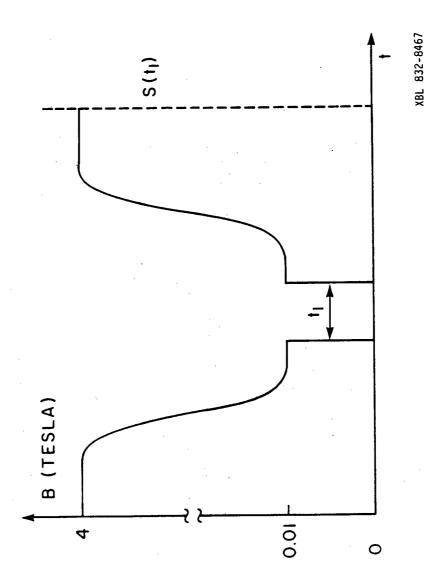
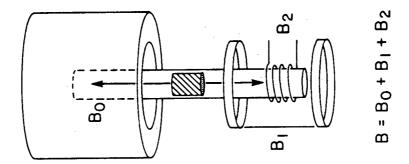
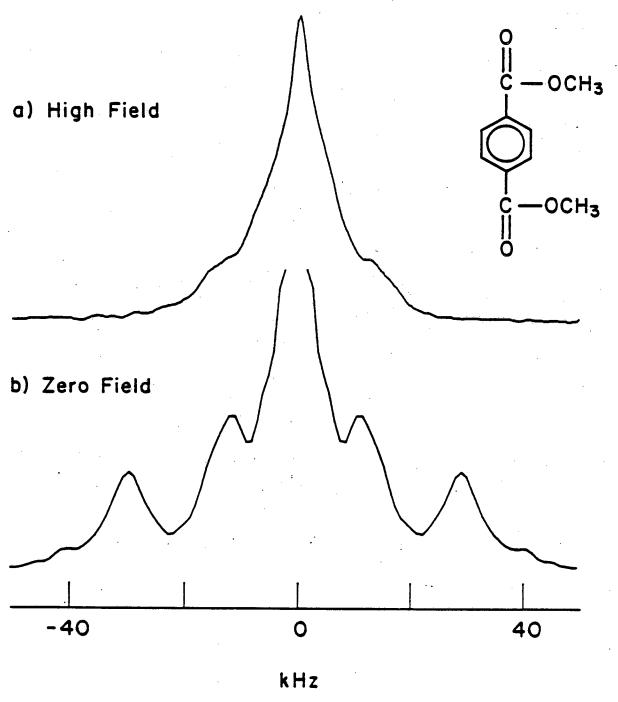


Figure 2





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Figure 3

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